

DENSITY PILLAR

MIDDLE SCHOOL LEVEL 1

Try your hand at this stackable liquid house of cards. This experiment is about fluids of various densities, studying their properties and how they interact. We will be using water-based liquids, meaning liquids made of water and other dissolved substances.

EDUCATIONAL STANDARDS:

NGSS CONNECTION:

MS-PS1-2. Analyze and interpret data on the properties of substances before and after the substances interact to determine if a chemical reaction has occurred.

COMMON CORE CONNECTION:

ELA/Literacy

RST.6-8.1 Cite specific textual evidence to support analysis of science and technical texts, attending to the precise details of explanations or descriptions.

RST.6-8.7 Integrate quantitative or technical information expressed in words in a text with a version of that information expressed visually (e.g., in a flowchart, diagram, model, graph, or table).
Mathematics

MP.2 Reason abstractly and quantitatively.

6.RP.A.3 Use ratio and rate reasoning to solve real-world and mathematical problems.

6.SP.B.4 Display numerical data in plots on a number line, including dot plots, histograms, and box plots.

6.SP.B.5 Summarize numerical data sets in relation to their context.

DOK:

Level 1: Recall

Level 2: Concept

Level 3: Strategic Thinking

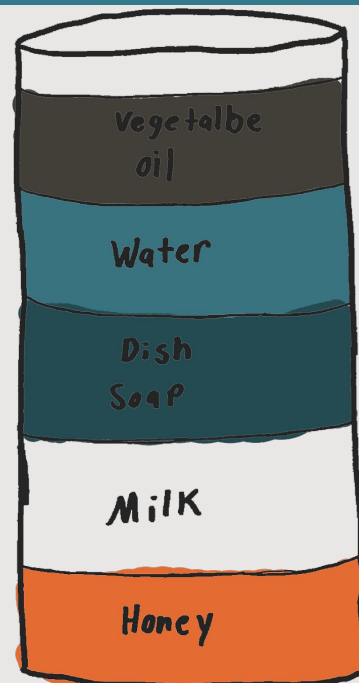
MATERIALS NEEDED:

- Water
- Food coloring
- Milk
- Honey
- Vegetable oil
- Dishwashing soap
- Tall glass
- Ping pong ball
- Cherry tomato
- Turkey baster

Try your hand at this stackable liquid house of cards. This experiment is about fluids of various densities, studying their properties and how they interact. We will be using water-based liquids, meaning made of water and and other dissolved substances.

DIRECTIONS:

1. Place your glass on a table and use the turkey baster to put the honey at the bottom of the glass.
2. Next add a layer of milk on top of the honey.
3. Add a layer of dish soap on top of the milk.
4. Add food coloring to the water and add a layer of that above the dish soap.
5. The final layer will be the vegetable oil, add that to the top.



Your pillar is built! Now, drop your ping pong ball on the top layer and watch, what happened? Try a cherry tomato next. Notice how different objects interact with the fluids differently.

OBJECTIVE:

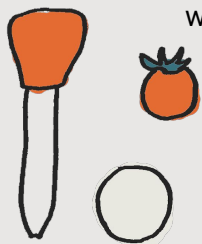
Students will be able to construct a density pillar and to interpret whether a new substance was formed.

ESSENTIAL QUESTIONS:

- How do substances interact when poured in a container?
- Do substances always mix?

ENGAGE:

1. Perform a density demo for students using oil and water.
 - a. Have two containers
 - i. One filled with water and food coloring
 - ii. One filled with oil
 - b. Have students rotate to the front of the room and record observations of the two liquids
 - c. Have students make predictions on what will happen when the two substances are “mixed”
 - d. Pour oil on top of the water
 - i. Have students record observations of the substances after they are mixed
 - ii. Have students make conclusion based on evidence on why the oil “floats” on the water
 - e. Repeat the process by pouring water into the oil
 - i. Make sure you have students make predictions, observations, and conclusions from their activity.



2. Evaluate
 - a. Informally students Predictions, observations, conclusions.

EXPLORE:

1. Provide students with the material list.
 - a. Have students observe each material independently
 - b. Students may record mass and volume of each substance
2. Ask students to perform experiments to determine what happens when various combinations of the substances are mixed
3. Evaluate
 - a. Process of data collection
 - b. Conclusions made from evidence

EXPLAIN:

1. Students and teacher should discuss why some substances float on others
2. Review of density with students may be required
3. Review of chemical and physical changes to determine if this applies here
4. These activities should be supplemented with other chemical and physical change experiments
 - a. See the Middle School Level 1 project: “Water Desalination”

ELABORATE:

1. Students use their data from above to construct a density pillar
2. Use their density pillar to discuss density and that the substances do not mix because of density.
3. If substances do mix discuss if a new substance was formed or if the molecules mixed is it a new substance?
4. Evaluate
 - a. Students use of data to draw conclusions
 - b. Reasoning